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25.1 Nuclear Radiation Define the

following terms:-radioactivity-

radiation-radioisotope-alpha particle-

beta particle- gamma ray 1. What

does the stability of a nucleus

depend on? 2. How does an

unstable nucleus release

energy? Chapter 25 Nuclear

Chemistry Study Guide

Answers The difference between

the mass number (the sum of the
numbers of protons and neutrons)

and the atomic number (the

number of protons) is equal to the

number of neutrons, so this nuclide

has 120 neutrons (201 - 81).

Atomic number = 81 mass number

= 201 81 protons 120

neutrons. Chapter 18 Nuclear

Chemistry Chapter 25 Nuclear

Answers

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